

Amendments to the Specification:

Please replace the paragraph beginning at page 15, line 3, with the following rewritten paragraph:

-- In some embodiments, higher amounts of iodide may be present in the photosensitive silver halide grains up to the saturation limit of iodide as described in U.S. Publication 2004/0053173 (Maskasky et al.) ~~eopending and commonly assigned U.S. Serial No. 10/246,265 (filed September 18, 2002 by Maskasky and Seacecia)~~. --

Please replace the paragraph beginning at page 19, line 10, with the following rewritten paragraph:

-- In addition, sulfur-containing compounds can be decomposed on silver halide grains in an oxidizing environment. Examples of such sulfur-containing compounds include sulfur-containing spectral sensitizing dyes described in U.S. Patent 5,891,615 (Winslow et al.) and diphenylphosphine sulfide compounds represented by the Structure (PS) described in U.S. Publication 2005/0123870 (Simpson et al.) ~~eopending and commonly assigned U.S.S.N. 10/731,251 (filed December 9, 2003 by Simpson, Burleva, and Sakizadeh)~~, both of which are incorporated herein by reference. --

Please replace the paragraph beginning at page 22, line 16, with the following rewritten paragraph:

-- Still other useful sources of non-photosensitive reducible silver ions in the practice of this invention are the silver core-shell compounds comprising a primary core comprising one or more photosensitive silver halides, or one or more non-photosensitive inorganic metal salts or non-silver containing organic salts, and a shell at least partially covering the primary core, wherein the shell comprises one or more non-photosensitive silver salts, each of which silver salts comprises a organic silver coordinating ligand. Such compounds are described in U.S. Patent 6,803,177 (Bokhonov et al.) ~~eopending and commonly assigned U.S. Serial No. 10/208,603 (filed July 30, 2002 by Bokhonov, Burleva, Whitecomb, Howlader, and Leichter)~~ that is incorporated herein by reference. --

Please replace the paragraph beginning at page 28, line 15, with the following rewritten paragraph:

-- Particularly useful toners are mercaptotriazoles as described in U.S. Patent 6,713,240 (Lynch et al.) ~~copending and commonly assigned U.S. Serial No. 10/193,443 (filed July 11, 2002 by Lynch, Zou, and Ulrich)~~, the heterocyclic disulfide compounds described in U.S. Patent 6,737,227 (Lynch et al.) ~~copending and commonly assigned U.S. Serial No. 10/384,244 (filed March 7, 2003 by Lynch and Ulrich)~~, the triazine-thione compounds described in U.S. Patent 6,703,191 (Lynch et al.), ~~copending and commonly assigned U.S. Serial No. 10/341,754 (filed January 14, 2003 by Lynch, Ulrich, and Skoug)~~. All of the above are incorporated herein by reference. --

Please replace the paragraph beginning at page 35, line 30, with the following rewritten paragraph:

-- The materials of this invention can include one or more antistatic agents in any of the layers on either or both sides of the support. Conductive components include soluble salts, evaporated metal layers, or ionic polymers as described in U.S. Patent 2,861,056 (Minsk) and U.S. Patent 3,206,312 (Sterman et al.), insoluble inorganic salts as described in U.S. Patent 3,428,451 (Trevoy), electroconductive underlayers as described in U.S. Patent 5,310,640 (Markin et al.), electronically-conductive metal antimonate particles as described in U.S. Patent 5,368,995 (Christian et al.), and electrically-conductive metal-containing particles dispersed in a polymeric binder as described in EP 0 678 776 A1 (Melpolder et al.). Particularly useful conductive particles are the non-acicular metal antimonate particles described in U.S. Patent 6,689,546 (LaBelle et al.) ~~copending and commonly assigned U.S. Serial No. 10/304,224 (filed on November 27, 2002 by LaBelle, Sakizadeh, Ludemann, Bhave, and Pham)~~. All of the above patents and patent applications are incorporated herein by reference. --

Please replace the paragraph beginning at page 36, line 18, with the following rewritten paragraph:

-- Additional conductive compositions include one or more fluorochemicals described in more detail in U.S. Patent 6,762,013 (Sakizadeh et al.) ~~copending and commonly assigned U.S. Serial No. 10/265,058 (filed October 4, 2002 by Sakizadeh, LaBelle, and Bhave)~~ that is incorporated herein by reference. --

Please replace the paragraph beginning at page 36, line 26, with the following rewritten paragraph:

-- Layers to reduce emissions from the film may also be present, including the polymeric barrier layers described in U.S. Patent 6,352,819 (Kenney et al.), U.S. Patent 6,352,820 (Bauer et al.), U.S. Patent 6,420,102 (Bauer et al.), and U.S. Patent 6,667,148 (Rao et al.), and U.S. Patent 6,746,831 (Hunt) ~~Serial No. 10/351,814 (filed January 27, 2003 by Hunt)~~, all incorporated herein by reference. --